Investigation on Dielectric and Optical Properties of Ba1-xCaxZr0.1Ti0.9O3 (x = 0.150) Ferroelectric Ceramics

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Abstract: Pellet samples of (Ba1-xCaxZr1-yTiy)O3 (x = 0.150, y = 0.90) were prepared using the solid-state reaction method following double sintering. The electrical and optical properties of (Ba1-xCaxZr1-yTiy)O3 (x = 0.150, y = 0.90) ceramics were studied. For (Ba0.85Ca0.15Zr0.1Ti0.9)O3 ferroelectric perovskite ceramics, the dielectric and structural properties were explored. The prepared composition of BCZT ceramics were calcined at 1100 °C. The sintering temperature was 1300 °C. Dielectric properties were observed from room temperature (RT) to 150 °C, at 1 MHz. The powder of the sintered samples has been taken for reflectance measurement at UV-Vis range. From the observed tauc plot, the band-gap is calculated. The measured band-gap is 3.19 eV.

Keywords- Dielectric constant; Electric conductivity; Reflectance; UV-Vis.

Introduction

Perovskite compounds are of considerable interest due to their high extent of tailorability in physical properties and application in the field of sensors. Their contributions in the field of sensors are significant due to their large piezoelectric properties. Ferroelectric/ABO3 type materials may generate voltage if external voltage or pressure has been applied to it. These compounds possess spontaneous polarization. Researchers investigated the dielectrics properties of (Ba1-xCaxZr1-yTiy)O3 (x = 0.150, y = 0.90) ceramics, near morphotropic phase boundary (MPB) region (Tian et al., 2013; Mondal et al., 2017; Hennings and Schnell, 1982; Yu et al., 2002; Wang et al., 2012; Wu et al., 2012). The BCZT ceramics show significant dielectric properties, at x = 0.150. At x = 0.150, Tian et al. observed co-existence of two different phases in BCZT ceramics (Tian et al., 2013). For (Ba0.85Ca0.15Zr0.1Ti0.9)O3 samples, O (“Orthorhombic”)- T (“Tetragonal”) transition point has been observed near 120 °C (Tian et al., 2013). The presence of these phases may create several possible polarization directions. Therefore, this MPB consisting of both phases and may have maximum possible polarization directions, which may contribute towards ks (coupling
A significant dielectric constant may be seen because of its large remnant polarization, which increases due to its large coupling factor (Yamashita and Shimanuki, 1996; Yamashita et al., 1996; Yamashita et al., 1998). In this study, the \((\text{Ba}_{1-x}\text{Ca}_x\text{Zr}_{1-y}\text{Ti}_y)\text{O}_3\) \((x = 0.150, y = 0.90)\) samples were prepared using the conventional “solid-state reaction” method with double sintering. The pellet samples were sintered at 1300 °C for 4 hrs. For dielectric measurement, the pellet samples were electroded with conducting silver paste, in M-I-M configuration. For XRD measurement, the prepared powder sample of BCZT composition was taken. Furthermore, PANalytical, X’PERT PRO X-ray machine has been used to measure XRD patterns. It has \(\text{CuK}_{\alpha1}\) radiation of wavelength 1.54060 Å. The range of 2θ angle was selected from 20 to 70°, at a scanning rate 0.3 sec per step with step size 0.03°.

Experimental details
The pellet samples of \((\text{Ba}_{1-x}\text{Ca}_x\text{Zr}_{1-y}\text{Ti}_y)\text{O}_3\) \((x = 0.150, y = 0.90)\) were prepared using solid-state reaction method with double sintering. Raw materials of Barium carbonate \((\text{BaCO}_3)\), Calcium carbonate \((\text{CaCO}_3)\), Zirconium oxide \((\text{ZrO}_2)\), and Titanium oxide \((\text{TiO}_2)\) were taken with purity 99.99%, were dried at 200 °C for 2 hours to remove the absorbed moisture. The methodology has been described in a previous communication (Biswas et al., 2020; Biswas et al., 2021; Biswas et al., 2019; Biswas et al., 2018; Biswas et al., 2021; Singh et al., 2018; Singh et al., 2019).

Result and discussion
X-ray Diffraction result
The XRD measurement was carried out for the \((\text{Ba}_{1-x}\text{Ca}_x\text{Zr}_{1-y}\text{Ti}_y)\text{O}_3\) \((x = 0.150, y = 0.90)\) composition, at the selected 2θ angle range. It has been shown in Fig. 1. For the prepared composition, powder sample was found in good consistency with “Inorganic Crystal Structure Database” (ICSD file No. 00-056-1033) data (Sen and Choudhary, 2004), with peaks corresponding to (020), (221), (402). The measured RT- XRD data of the \((\text{Ba}_{0.85}\text{Ca}_{0.15}\text{Zr}_{0.1}\text{Ti}_{0.9})\text{O}_3\) powder samples exhibit orthorhombic phase. The lattice parameters were measured manually. The selected space group is ‘C’ type. For the prepared powder samples, the measured X-ray diffraction data were matched with “Inorganic Crystal Structure Database” (ICSD) data. The ICSD data will give the information of the orthorhombic unit cell \((a, b\text{ and } c\text{ parameters})\) and volume. The manually calculated values \((a, b\text{ and } c\text{ parameters})\) must be in more close approximation to those values. At first, the positions of the observed x-ray peaks were
identified. For a corresponding peak, the d-spacing and (hkl) has been noted from the ICSD data. The equation for an interplaner spacing for an orthorhombic unit cell was used to calculate the lattice parameters. To work with the above interplaner equation, the h, k, l, and d values were noted and further, the lattice parameters were measured. The lattice parameters were given in Table 1.

Table 1. Lattice parameters of (Ba_{1-x}Ca_{x}Zr_{1-y}Ti_{y})O_{3} (x = 0.150, y = 0.90) samples and peaks used in the present calculation were (111), (020) and (402).

<table>
<thead>
<tr>
<th>Cell Constants (Calculated)</th>
</tr>
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<tbody>
<tr>
<td>(Ba_{0.85}Ca_{0.15}Zr_{0.1}Ti_{0.9})O_{3}</td>
</tr>
<tr>
<td>0.150</td>
</tr>
</tbody>
</table>

![XRD pattern of (Ba_{1-x}Ca_{x}Zr_{1-y}Ti_{y})O_{3} (x = 0.150, y = 0.90) ceramics, at room temperature.](image)

**Fig. 1.** XRD patterns of (Ba_{1-x}Ca_{x}Zr_{1-y}Ti_{y})O_{3} (x = 0.150, y = 0.90) ceramics, at room temperature.

**Dielectric and morphology results**

In this study, the relative permittivity (K), loss tangent (tan δ), and dielectric conductivity (σ) were observed systematically. For the prepared samples, K, tan δ, and σ were measured from temperature 30 and 140 °C, at 1 MHz (Figs. 2-3). The micro-structure show fine grain and homogeneous structure of the (Ba_{0.85}Ca_{0.15}Zr_{0.1}Ti_{0.9})O_{3} samples. The grain size was observed between 0.7 and 1.0 μm (Fig. 4). The average grain size was measured 1.15 μm. The dielectric constant was found maximum near 120 °C.
Furthermore, the dielectric constant has been decreasing with an increase in temperature. The large value of the dielectric constant indicates a phase change near 120 °C. The transition is also observed from the data of electric conductivity. The electric conductivity was suddenly increased near 120 °C and further, it decreases. The loss of the prepared composition is very small at high frequency.

Fig. 2. Temperature-dependent dielectric constant and loss tangent of the (Ba_{1-x}Ca_xZr_{1-y}Ti_y)O_3 (x = 0.150, y = 0.90) sample, at 1 MHz.
**Fig. 3.** Temperature-dependent electric conductivity of the (Ba$_{1-x}$Ca$_x$Zr$_{1-y}$Ti$_y$)O$_3$ (x = 0.150, y = 0.90) ceramics, at 1 MHz.

![Fig. 3](image1)

**Fig. 4.** SEM images of (Ba$_{1-x}$Ca$_x$Zr$_{1-y}$Ti$_y$)O$_3$ (x = 0.150, y = 0.90) ceramics, at different magnification.

**Table 2:** Energy dispersive X-ray results of the (Ba$_{1-x}$Ca$_x$Zr$_{1-y}$Ti$_y$)O$_3$ (x = 0.150, y = 0.90) system and spectrum analysis.

<table>
<thead>
<tr>
<th>ELEMENT</th>
<th>WEIGHT%</th>
<th>ATOMIC%</th>
</tr>
</thead>
<tbody>
<tr>
<td>O K</td>
<td>25.69</td>
<td>66.10</td>
</tr>
<tr>
<td>Ca K</td>
<td>2.39</td>
<td>2.46</td>
</tr>
<tr>
<td>Ti K</td>
<td>16.36</td>
<td>14.06</td>
</tr>
<tr>
<td>Zr L</td>
<td>4.81</td>
<td>2.17</td>
</tr>
<tr>
<td>Ba L</td>
<td>50.75</td>
<td>15.21</td>
</tr>
<tr>
<td>Totals</td>
<td>100.00</td>
<td></td>
</tr>
</tbody>
</table>

**Table 3.** Standards selected in EDAX analysis of the (Ba$_{1-x}$Ca$_x$Zr$_{1-y}$Ti$_y$)O$_3$ (x = 0.150, y = 0.90) ceramics.

<table>
<thead>
<tr>
<th>STANDARD DETAILS</th>
</tr>
</thead>
<tbody>
<tr>
<td>O SiO$_2$</td>
</tr>
<tr>
<td>Ca Wollastonite</td>
</tr>
<tr>
<td>Ti Ti</td>
</tr>
<tr>
<td>Zr Zr</td>
</tr>
<tr>
<td>Ba BaF$_2$</td>
</tr>
</tbody>
</table>
Lossy dielectric can be characterized with the help of a circuit analog of a resistance in parallel with a capacitor (Haertling, 1967). When the input signal is taken as a sinusoidal signal, the capacitor always show low reactance path at higher frequencies, leading to minimization in conduction losses. Therefore, dielectric loss has been decreasing with an increase in frequency, which is consistent with the previous reports (Haertling, 1967; Jaeger and Egerton, 1962). The EDAX results confirmed the atomic percentage of the elements. The data has been given in Table 2. From the spectrum of EDAX, the presence of the elements can be confirmed. The standard details were given below in Table 3.

**Reflectance results**

The optical properties of the prepared composition have been observed (Figs. 5-6). The band-gap was measured 3.19 eV (Tunc et al., 2019). The band-gap is calculated using tauc function. Fig. 4 indicates the reflectance graph with the variation in wavelength. The reflectance is 75% at around 900 nm (UV range) and 10% at around 350 nm (Vis range). A sharp downfall may be observed in reflectance plot at around 350 nm. Teflon is taken as the standard material.

![Reflectance graph](image)

**Fig. 5.** Variation in reflectance with the wavelength, in $(\text{Ba}_{1-x}\text{Ca}_x\text{Zr}_{1-y}\text{Ti}_y)\text{O}_3$ $(x = 0.150, y = 0.90)$ ceramics, at 1 MHz.
Concentration

The dielectric and optical properties of the ferroelectric perovskite (Ba$_{0.85}$Ca$_{0.15}$Zr$_{0.1}$Ti$_{0.9}$)O$_3$ ceramics were observed. The composition shows polycrystalline structure. The observed maximum dielectric constant is measured 729, at around 100 °C. Also grain size is in µm range. From the observed optical properties, the reflectance was observed 80% near UV wavelength and 10% near visible range. The band-gap of the prepared composition is 3.19 eV.

References

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Fig. 6. Tauc plot of the (Ba$_{1-x}$Ca$_x$Zr$_{0.1}$Ti$_{0.9}$)O$_3$ (x = 0.150, y = 0.90) powder samples.
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